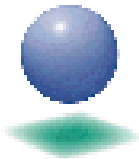




# **Sustainable Approaches to Meeting Potential Future Nutrient Limits – Part 2**

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CH2M HILL

OWEA Annual Conference  
23 June 2011



**CH2MHILL**

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# Presentation Outline

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- Why Remove Nutrients?
  - Overview of Wastewater Treatment
  - Nutrient Removal
    - Phosphorus Removal
    - Nitrogen Removal
  - Sustainability Perspective
  - Design & Operational Considerations
  - Take Home Messages
- Part 1
- Part 2

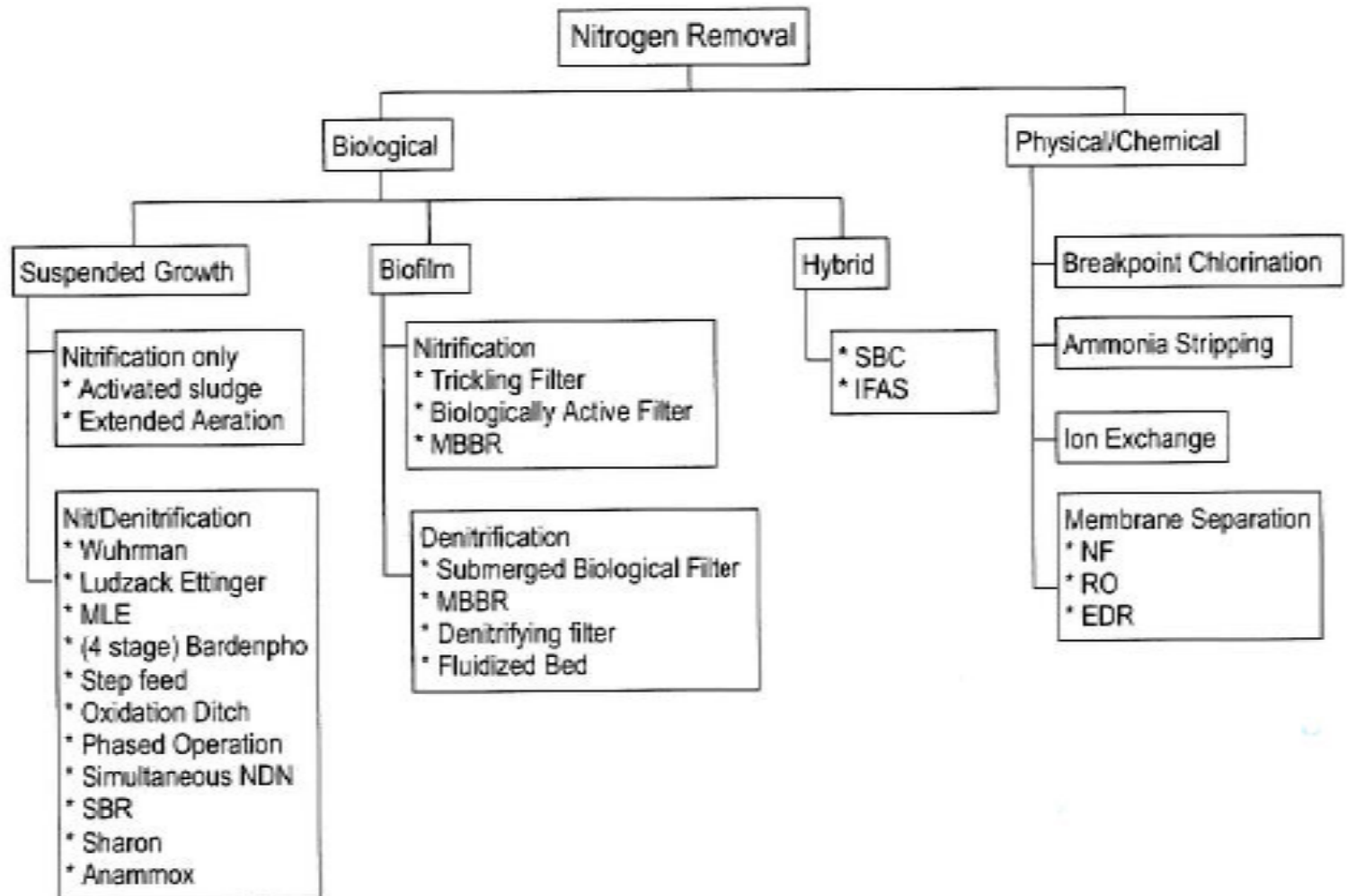


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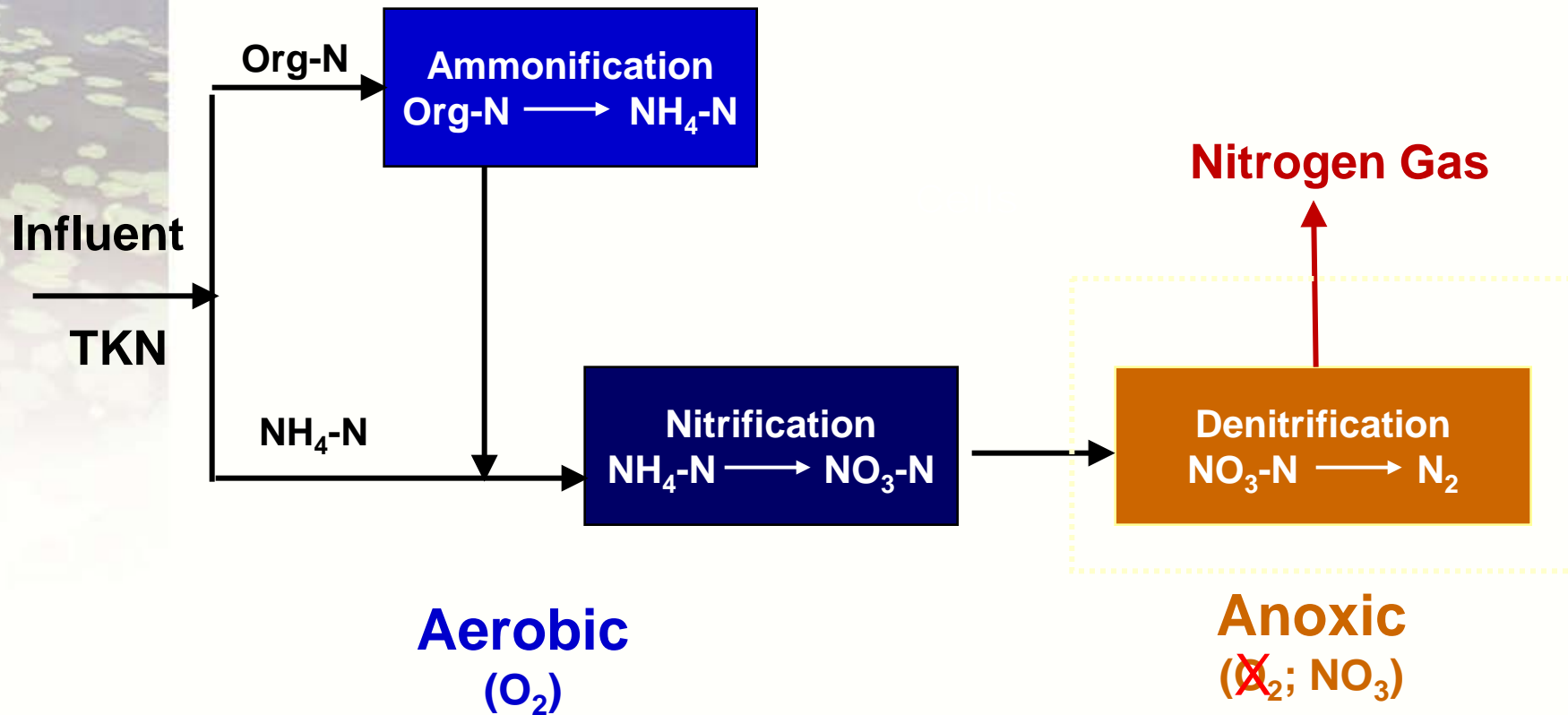
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- Why Remove Nutrients?
- Overview of Wastewater Treatment
- **Nutrient Removal**
  - Phosphorus Removal
  - **Nitrogen Removal**
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# Nitrogen Removal Alternatives



# Nitrogen Removal Processes





# Nitrification

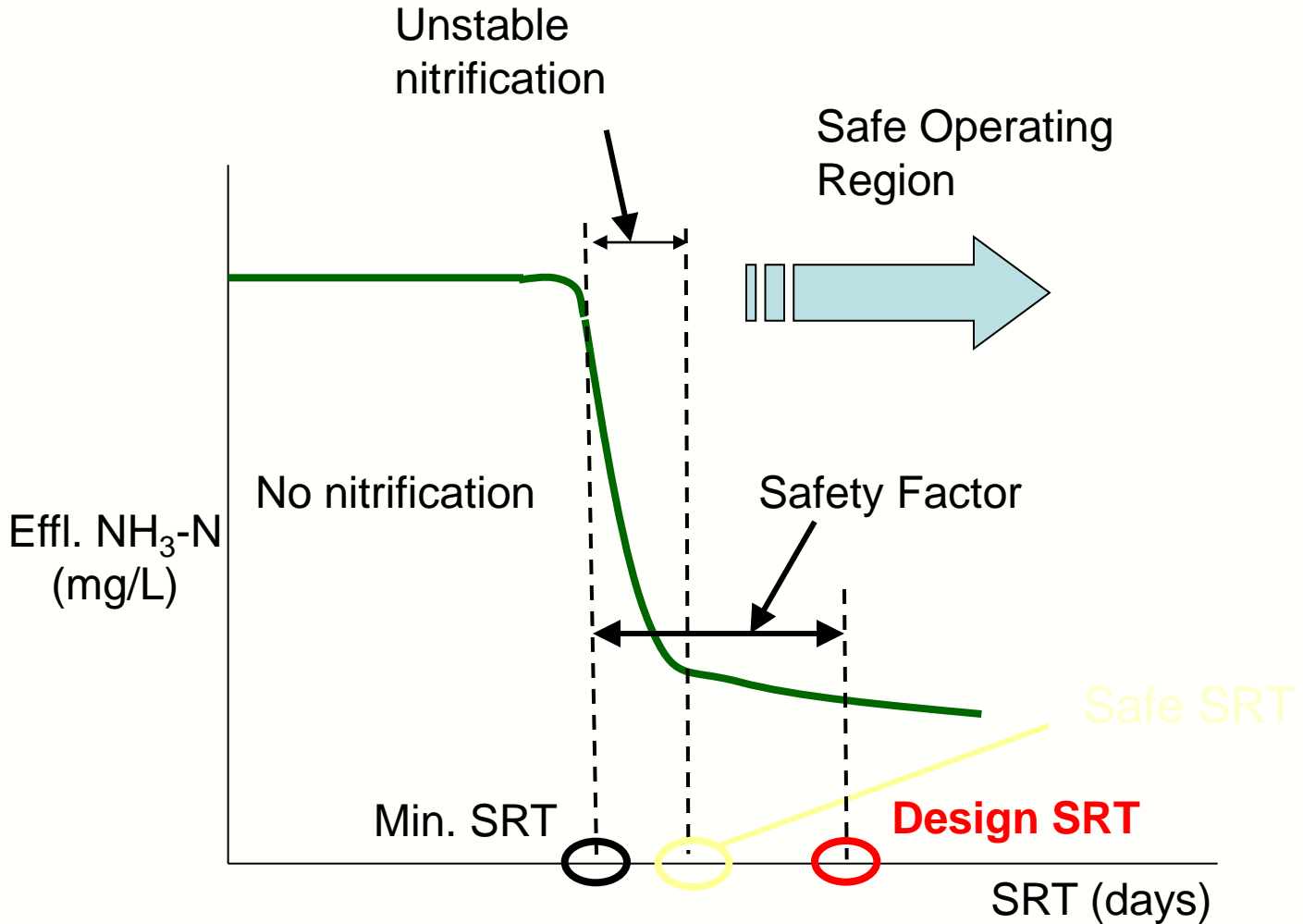
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- Mediated by *Autotrophs*
- Conversion (oxidation) of ammonia to nitrate
- Does not result in significant nitrogen removal

## Factors that impact nitrification

- Solids Retention Time (SRT)
- pH (Optimum: 6.5 - 8.0)
- DO (Optimum: 1.0 – 2.0 mg/L)
- Absence of inhibitory compounds
- Recycle load

# Adequate SRT is Crucial for Reliable Nitrification



# Denitrification

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- Denitrification: Conversion (reduction) of nitrate to nitrogen gas



- Facultative heterotrophic organisms
- Results in nitrogen removal

## Factors impacting denitrification

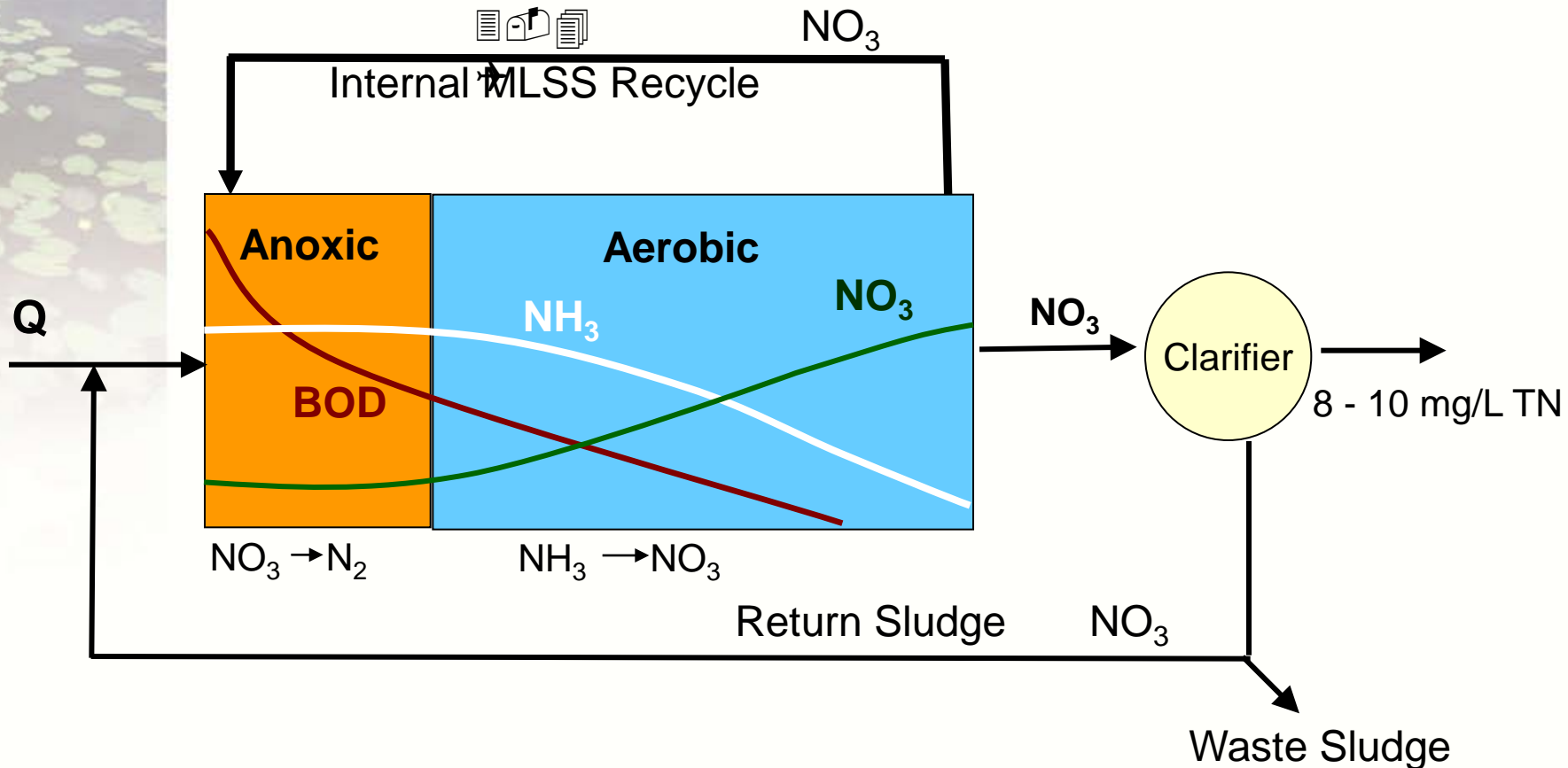
- Anoxic condition (Sufficient  $\text{NO}_3$ )
- Carbon source
- Adequate anoxic SRT
- Temperature

# Nitrification vs. Denitrification

Factor	Nitrification	Denitrification
Environment	Aerobic	Anoxic
Oxygen source (Electron Acceptor)	Dissolved oxygen (DO)	Combined Oxygen (NO <sub>3</sub> )
Type of Organism	Autotrophs	Heterotrophs
Energy (food) source Electron Donor	Ammonia-N	Organic Matter
Carbon Source	Organic Matter	Inorganic (CO <sub>2</sub> )
Oxygen	Demand (4.6 lb O <sub>2</sub> / lb NH <sub>4</sub> oxidized)	Credit (2.9 lb O <sub>2</sub> / lb NO <sub>3</sub> reduced)
Alkalinity (as CaCO <sub>3</sub> )	Consumed (7.1 lb /lb NH <sub>4</sub> oxidized)	Produced (3.6 lb /lb NO <sub>3</sub> reduced)

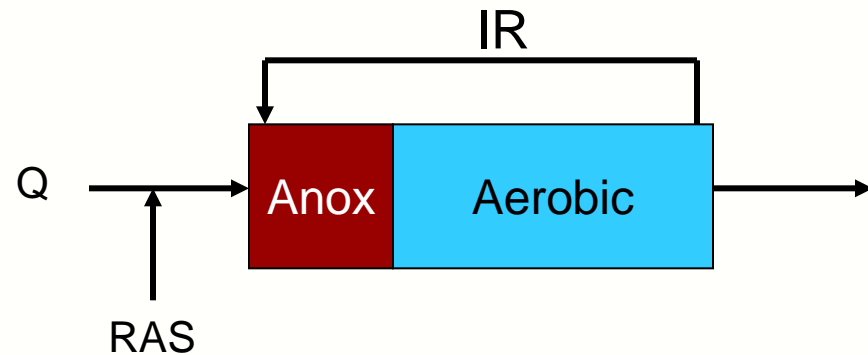
# How is Nitrogen Removal Achieved in Practice?

## Modified Ludzack-Ettinger (MLE) Process

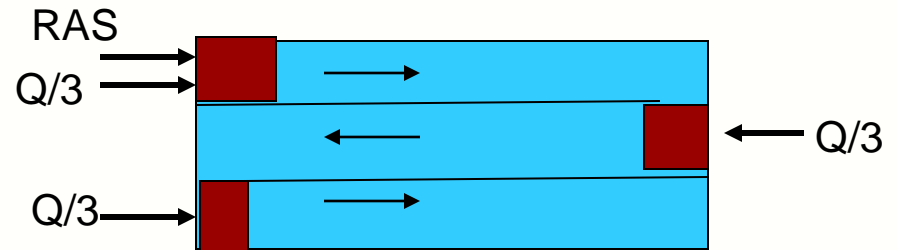


# Nitrogen Removal Capability of Various Processes

MLE Process  
TN = 8 - 10 mg/L

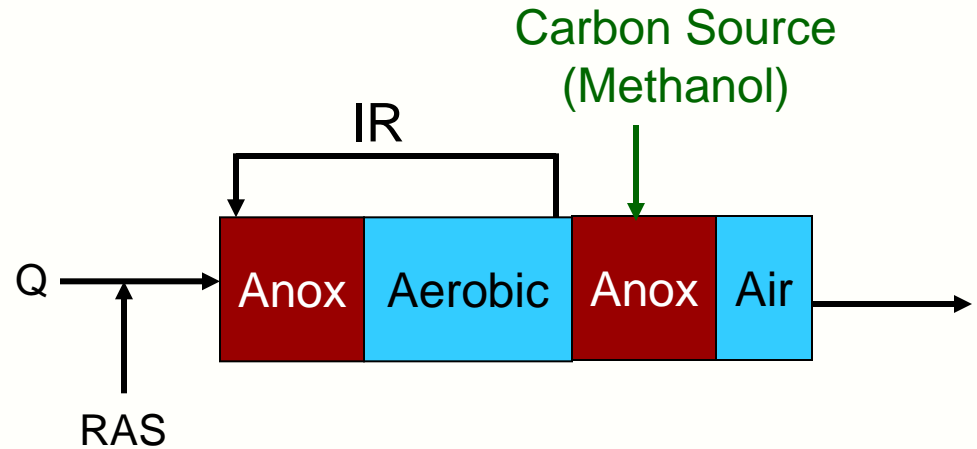


Step-Feed  
TN = 8-10 mg/L

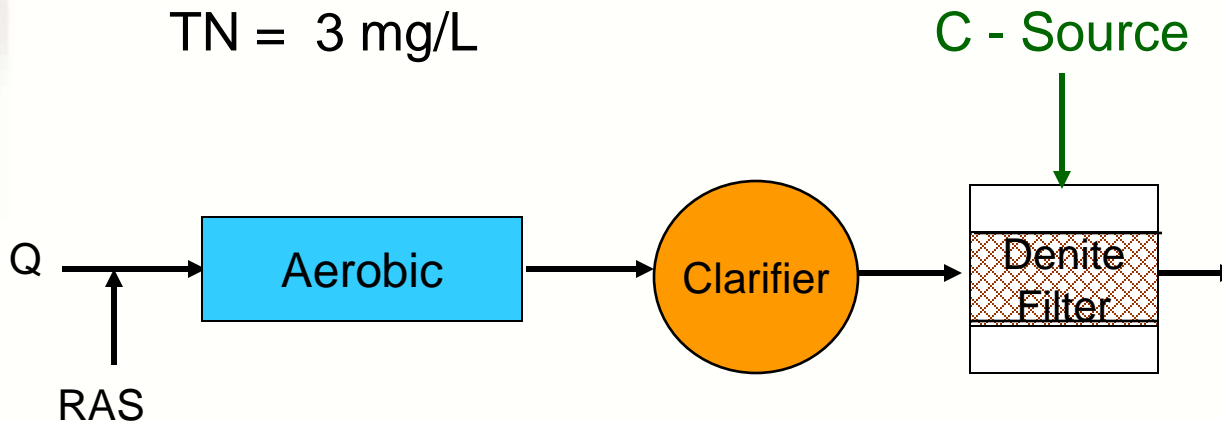


# Nitrogen Removal Capability of Various Processes

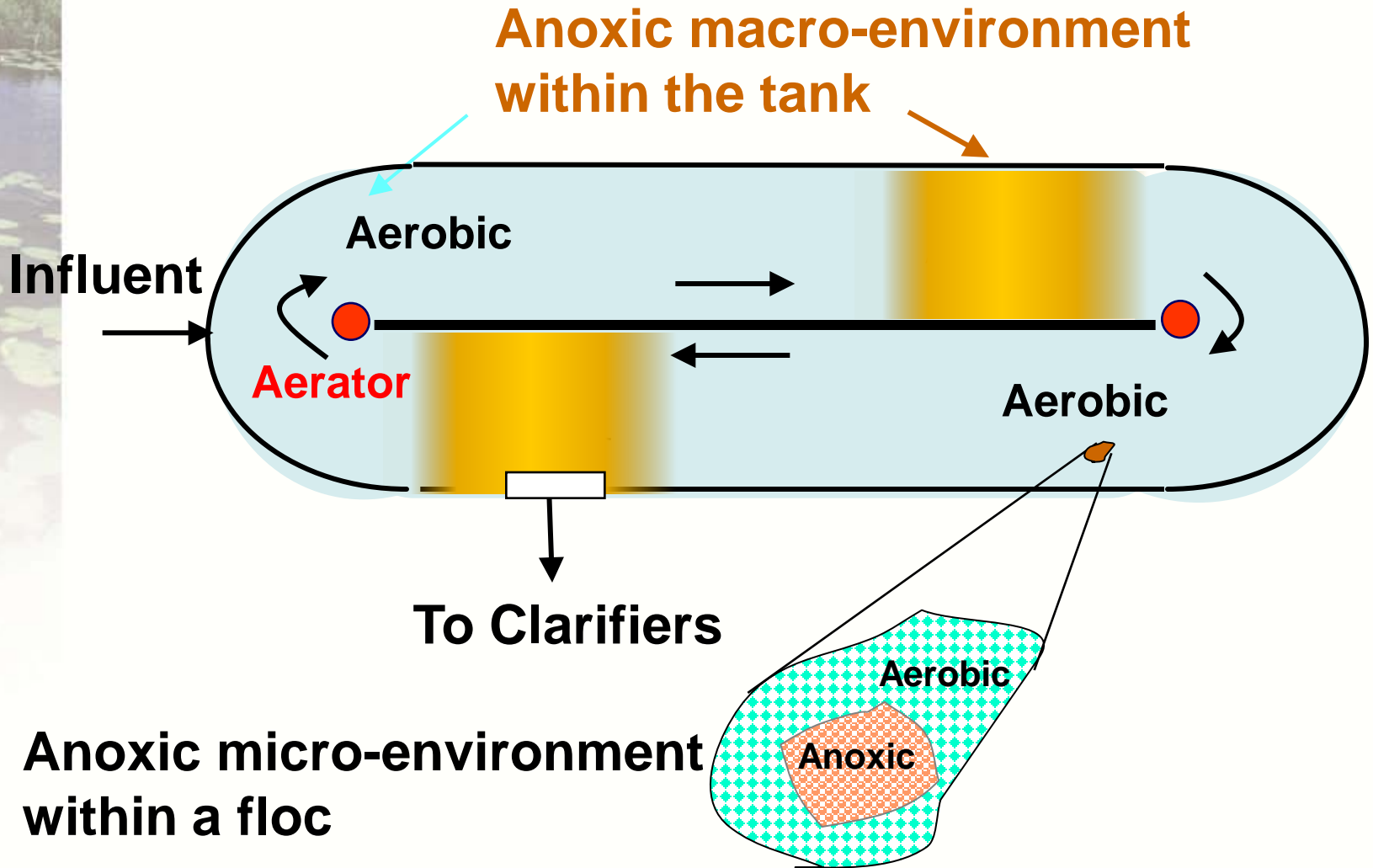
4-Stage Bardenpho  
TN = 3 mg/L



Denitrification Filter  
TN = 3 mg/L



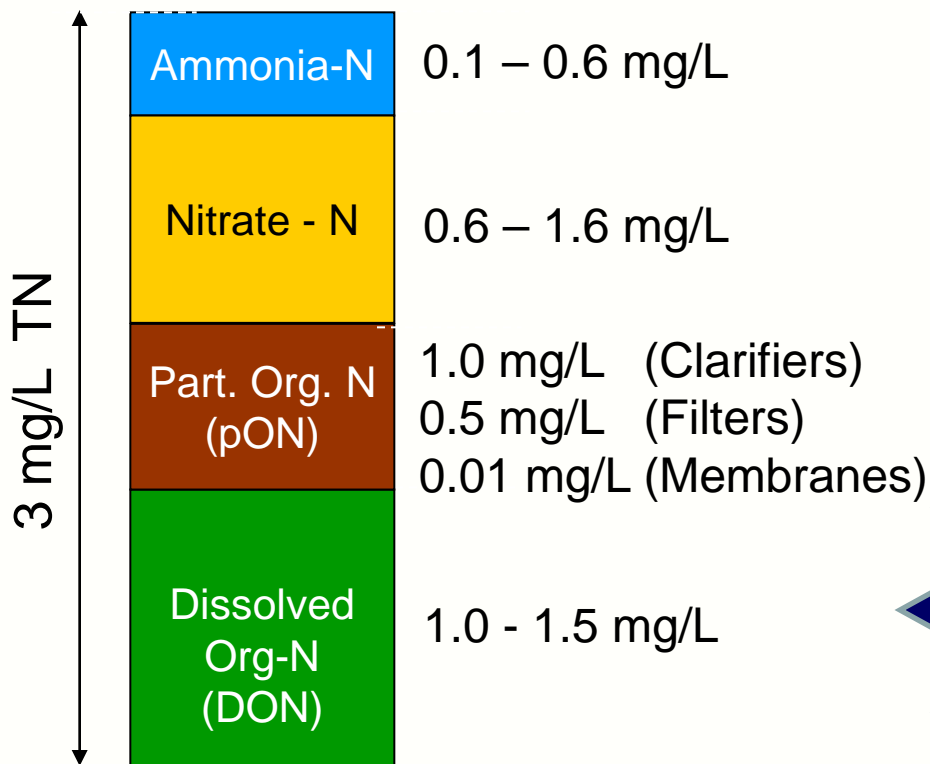
# Informal Anoxic Zones Cause Simultaneous Nitrification/Denitrification



**Anoxic micro-environment within a floc**

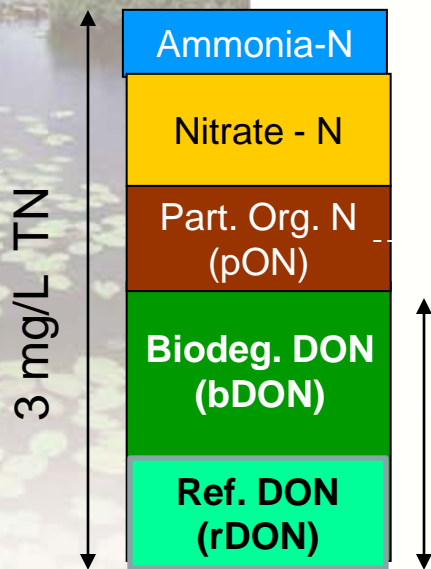
# What is the Limit Of Technology (LOT) for TN

- The LOT with conventional treatment processes is 3 mg/L TN



Can we  
go lower  
than  
LOT?

# Achieving $< 3$ mg/L TN Requires an Understanding of TN Speciation



- DON is the largest TN fraction of effluent (up to 50%)
- Achieving  $< 3$ mg/L (LOT) will require DON removal
- DON consists of :
  - Biodegradable DON (bDON)
  - Refractory DON (rDON)

## Potential DON removal technologies

Process	bDON	rDON
Nitrification	+++	-
Denitrification w/supplemental C	+	-
Advanced oxidation	+	+
Reverse Osmosis	+++++	+++++



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# Striking a Balance

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- Nutrient removal facilities:
  - Incur significant capital and O&M costs
  - Have larger carbon footprint
    - Increased use of chemicals & energy – increased GHG emissions

A decision to implement nutrient removal should be based on 'net' environmental benefit & costs

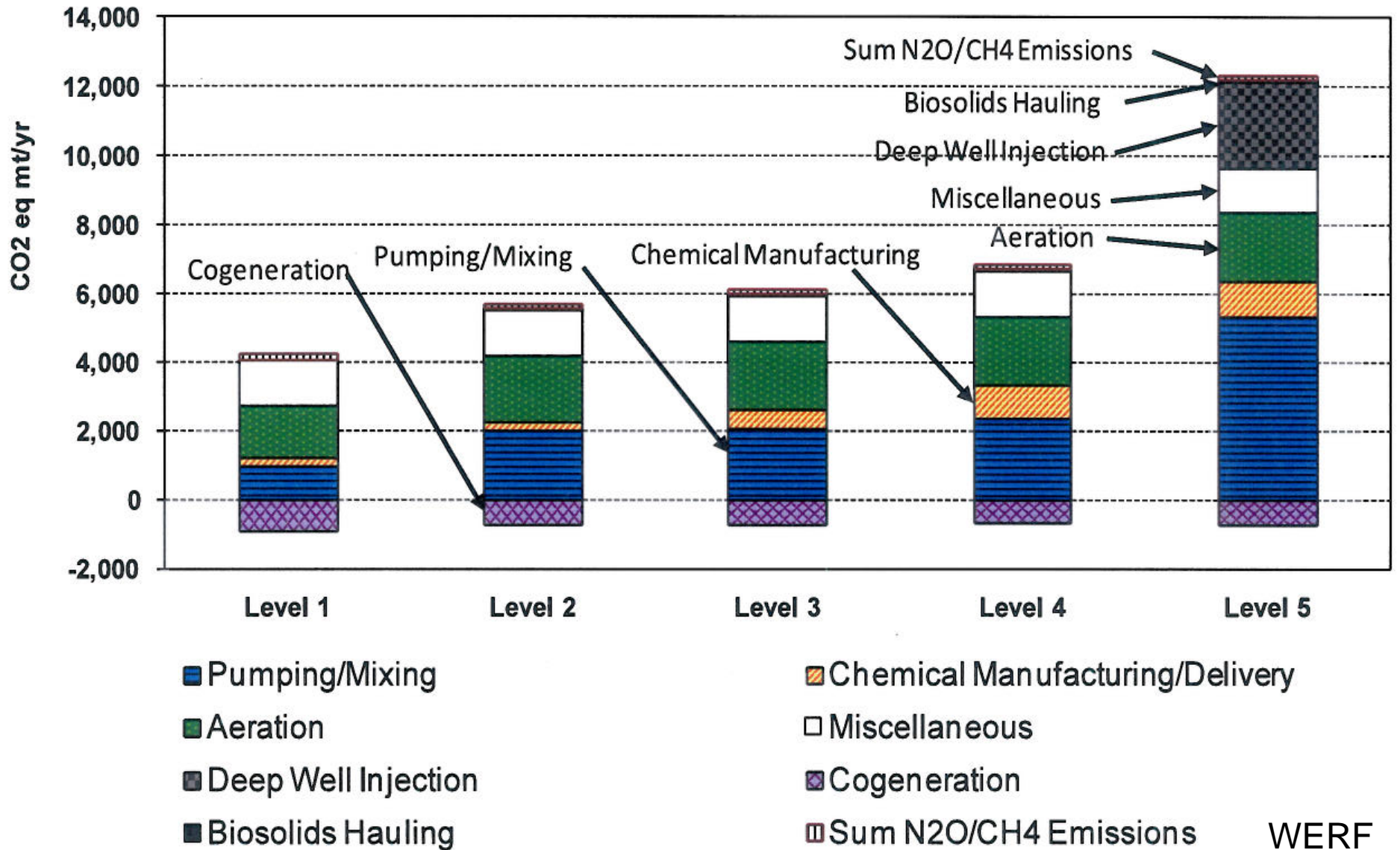
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# Summary of Ongoing WERF Study

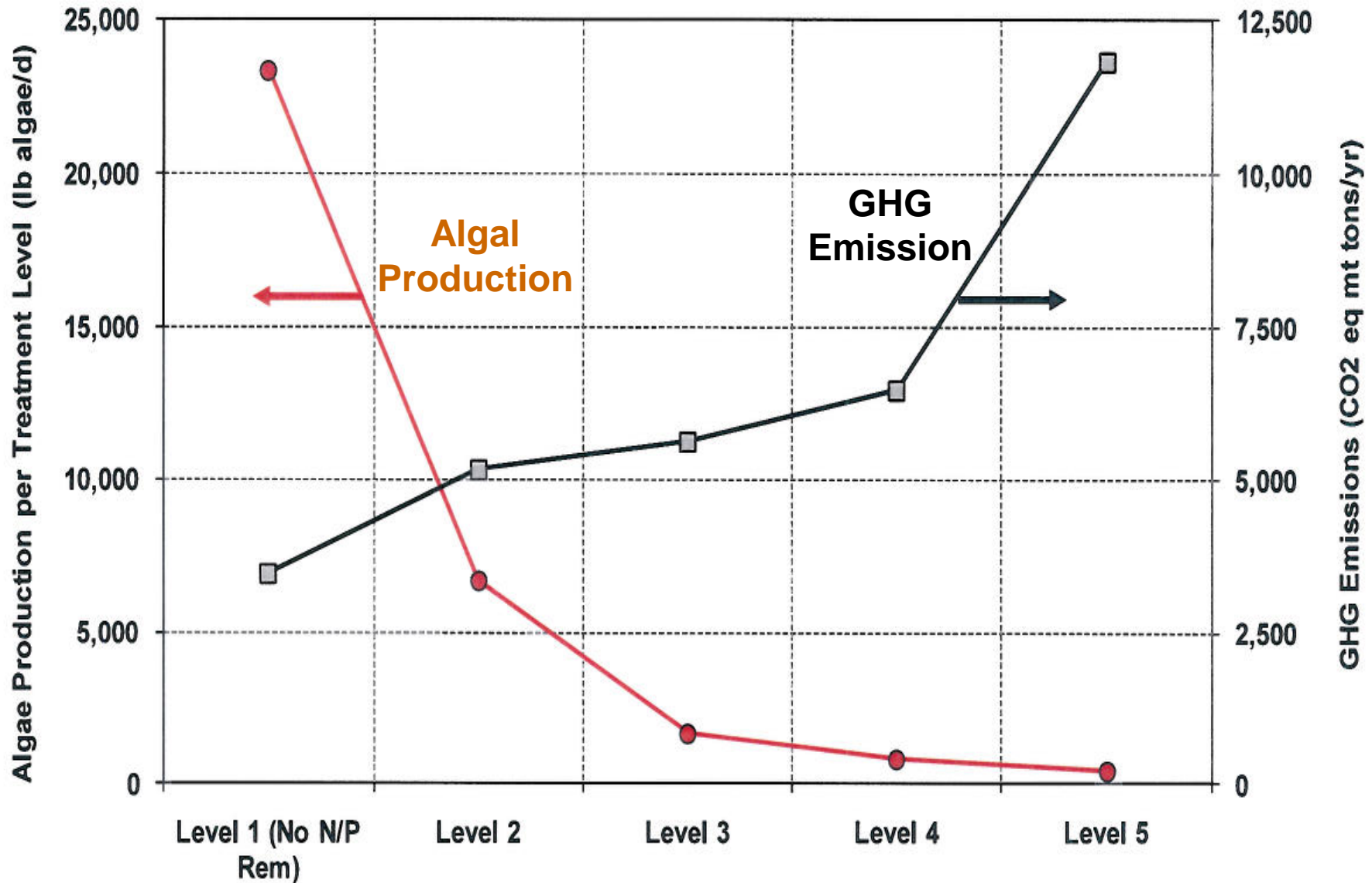
- 10 mgd plant with following effluent objectives

Level	BOD (mg/L)	TSS (mg/L)	TN (mg/L)	TP (mg/L)
1 (Secondary)	30	30	-	-
2	-	-	8	1
3	-	-	4-8	0.1 – 0.3
4	-	-	3	0.1
5	-	-	<2	<0.05

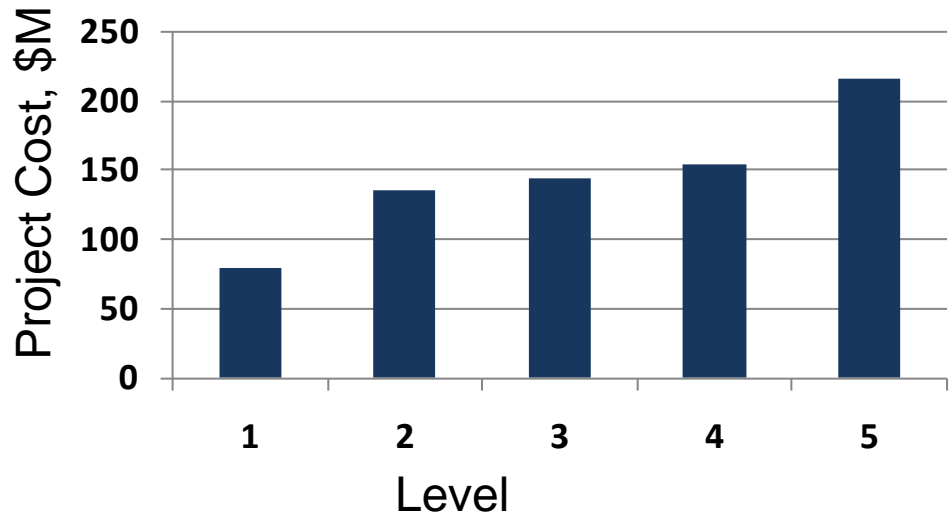
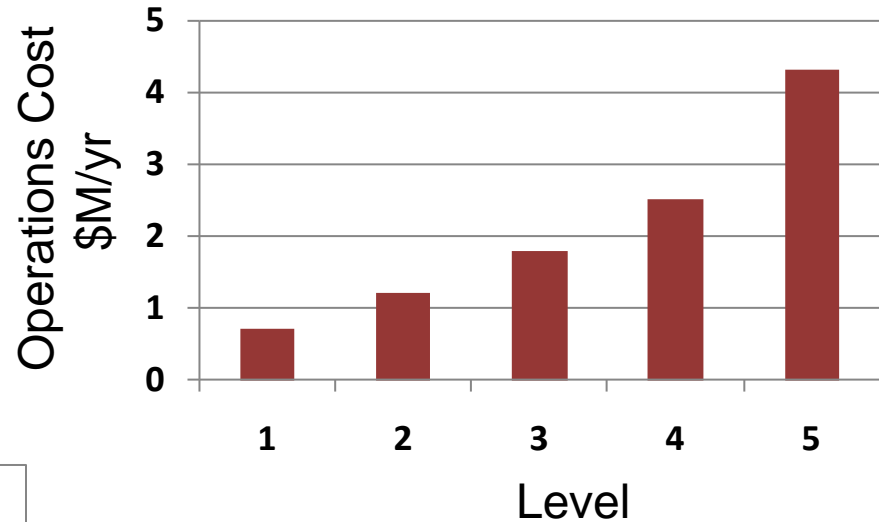
# Total GHG Emission Burden



# GHGs versus Algal Production



# Capital & Operations Costs



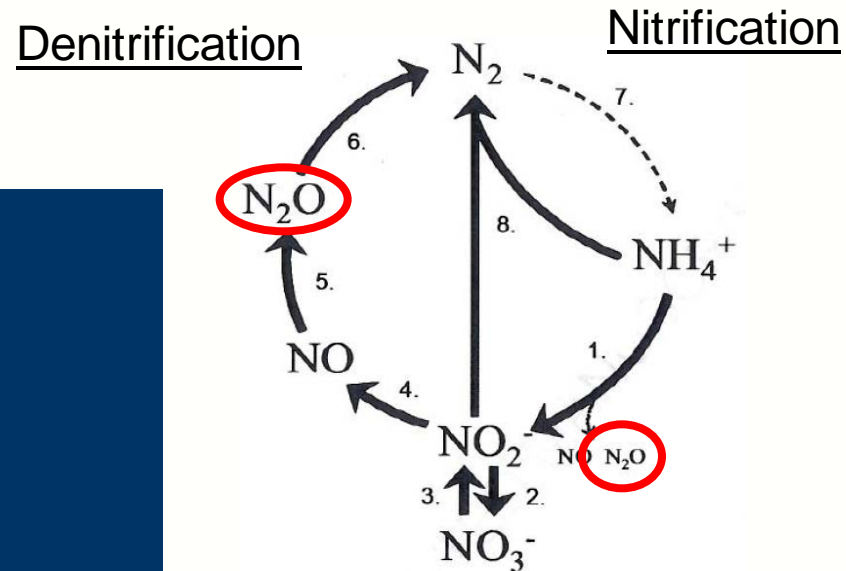
Excludes labor, maintenance, & membrane replacement costs

WERF

# New Information on Nitrous Oxide Generation at WWTPs



- $\text{N}_2\text{O}$  potentially produced during nitrification & always during denitrification.
- Potentially more during nitrification. Occurs due to:
  - Intermittent or incomplete nitrification
  - Repeated switching between aerobic & anoxic + high DO &  $\text{NH}_3\text{-N}$
- But lower  $\text{N}_2\text{O}$  production in ox. ditches due to more uniform spatial DO profile



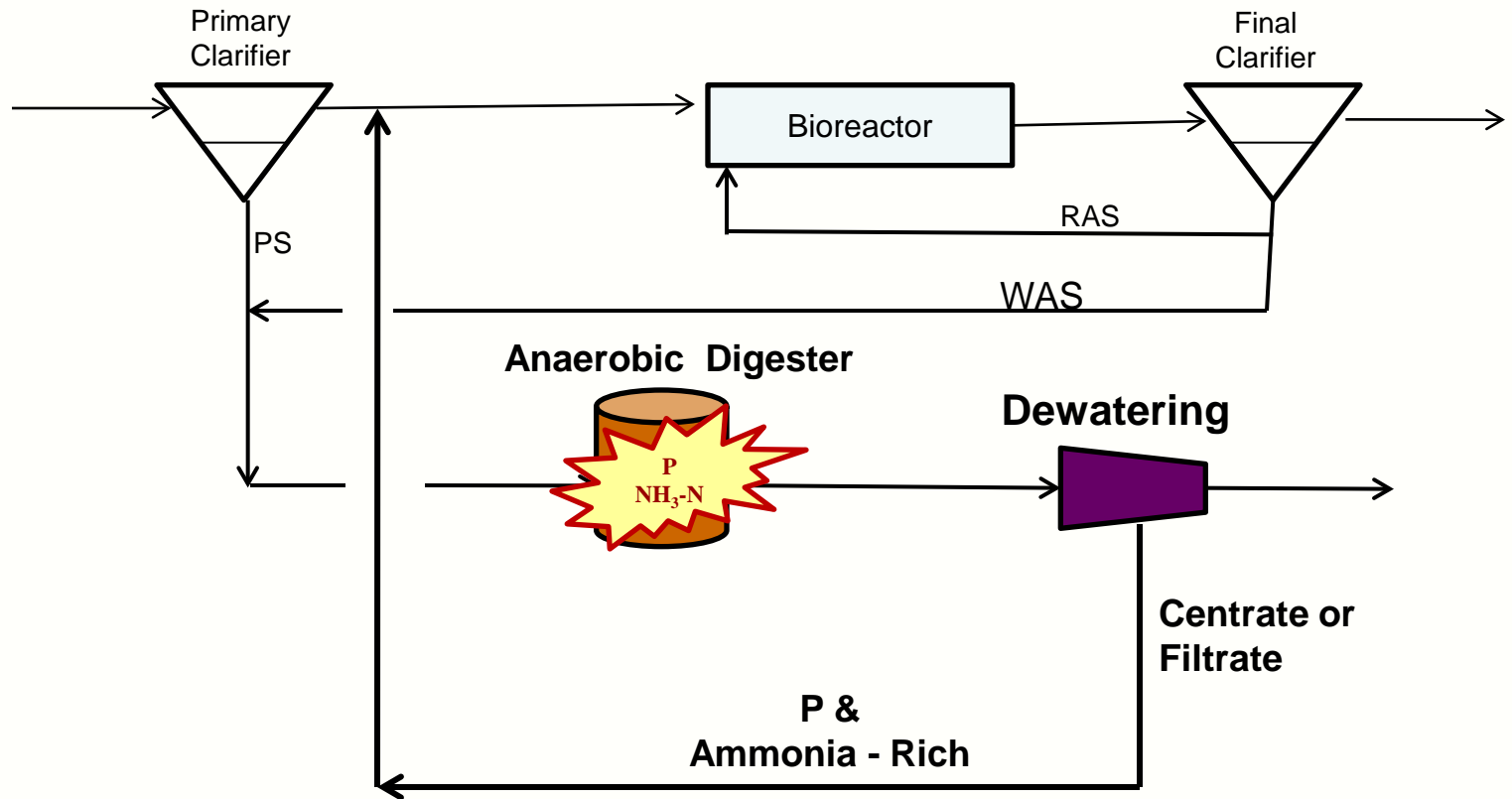


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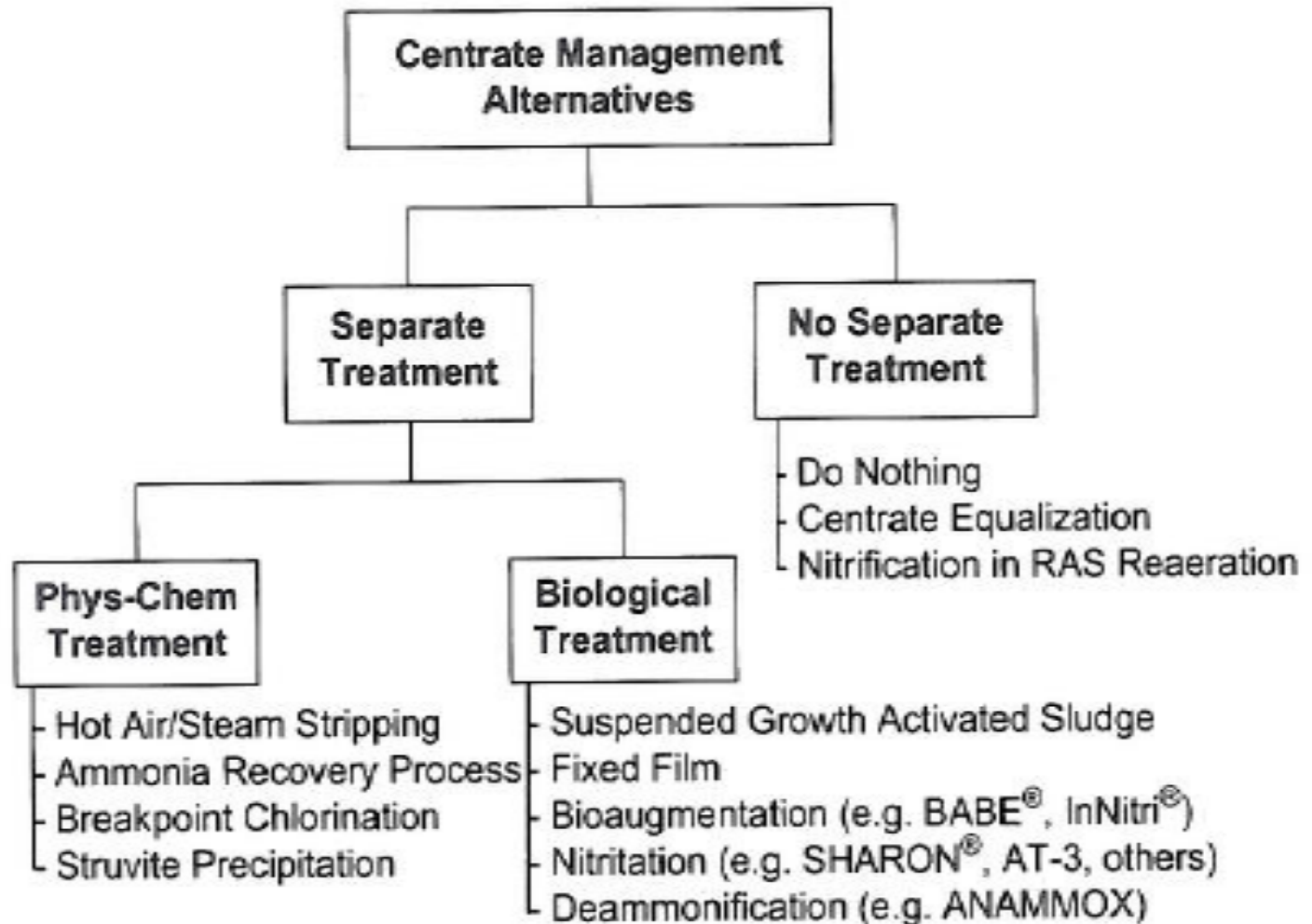
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# Recycle Management is Crucial for Meeting TP & TN Limits



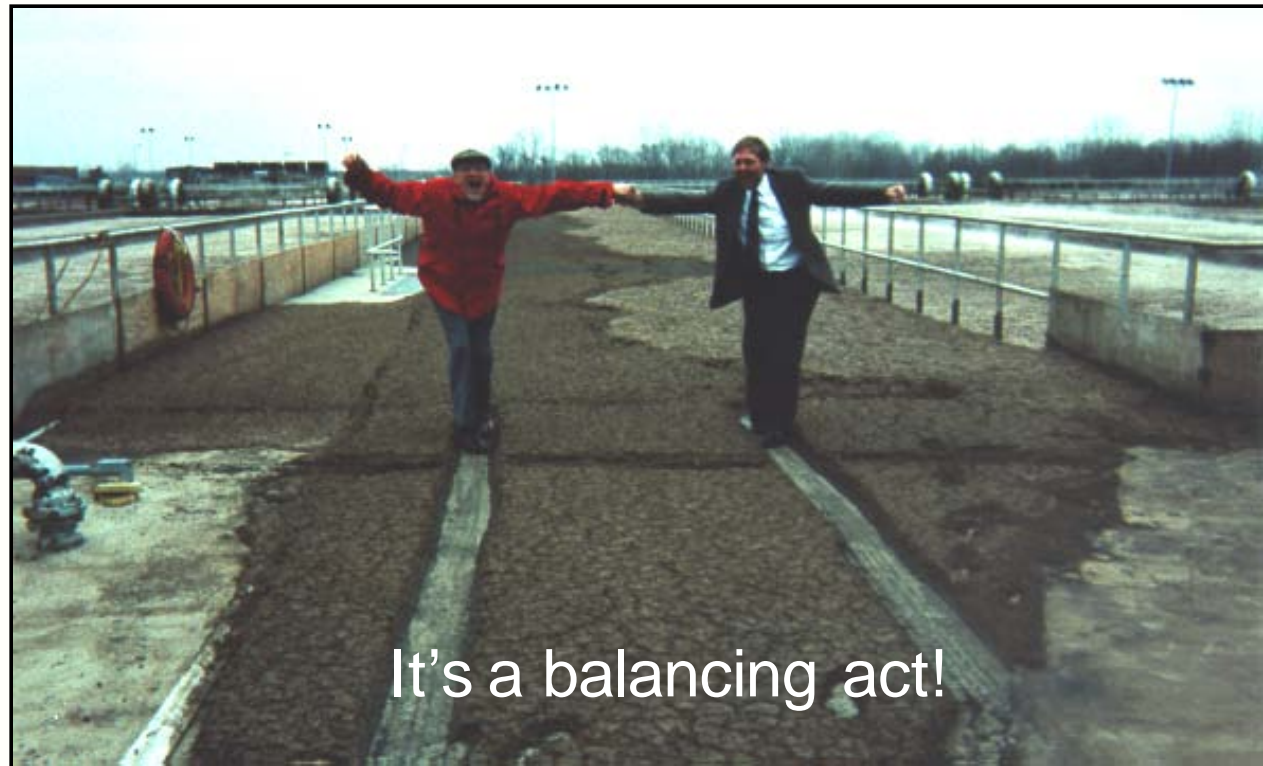
- Anaerobic digestion releases P & ammonia.
- Dewatering centrate/filtrate can impose additional load on the main process

# Sidestream Management Alternatives



# The Dreaded Foam

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It's a balancing act!

# More Design & Operational Considerations

- General design approach:

- Influent characteristics
- Effluent limits
- The box in-between



- General process optimization approach:

- First, nitrification & denitrification
- Then, EBPR; add supplemental chemical, if required

- Smart automation:

- Judicious use of instrumentation
  - Aeration control
  - SRT control

# More Design & Operational Considerations

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- Even flow splitting
  - Underloaded units often can not compensate for overloaded units
- Provision to control bulking & foaming.
- It takes two to tango!
  - Designers: provide operational flexibility.
  - Operators: use operational flexibility provided.

# Take Home Messages

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- Technology limits:
  - One anoxic zone/Ox ditches: 8 – 10 mg/L TN
  - Two anoxic zones/Denite filter: 3 mg/L TN
  - Tertiary treatment to remove DON: <3.0 mg/L TN
- The decision to achieve low N & P should be based on sustainability analysis.
- Data is not information.
- OH plants are not likely to see TN limit in the immediate future.



# What Will the Ideal Plant of the Future Look Like?

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Characteristics of the ideal plant of the future:

1. TN = 3-5 mg/L
2. TP  $\leq$  0.1 mg/L
3. Energy neutral
4. Resource recovery (P recovery)
5. Wet weather flow processing
6. Effluent reuse

Incorporate provision so that a plant can gradually evolve into the ideal plant of the future in a cost effective manner.

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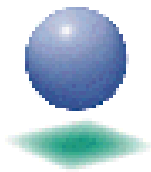
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